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The mass of substance A is 6.15 g and its initial temperature is 20.5 C. The mass of substance B is 25.2 g and its initial

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temperature is 52.7 C. The final temperature of both substances at thermal equilibrium is 46.7 C. If the specific heat capacity of substance B is 1.17 J/g # C , what is the specific heat capacity of substance A?

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Sodium hypochlorite - Wikipedia

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Ground State Electron Configuration: Definition & Example

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If a sample of ammonia decomposes in a closed system and the concentration of N_2 increases by 0.11 M, the change in the N_2 concentration, $\Delta[N_2]$, the final concentration minus the initial concentration, is 0.11 M. The change is positive because the

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concentration of N_2 increases.. The change in the H_2 concentration, $\Delta[H_2]$, is also positive—the concentration of H_2 increases as ammonia ...

13.4 Equilibrium Calculations - Chemistry

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Properties of Liquids - Chemistry

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Thermochemistry Heat capacity (C) 218.6 J/(K·mol) Std molar ...
pure β - D- glucose of +17.5°·ml/(dm·g). When equilibrium has
been reached after a certain time due to mutarotation, the angle
of rotation is +52.7°·ml/ ... Another test is the 2-hour glucose
tolerance test (GTT) - for this test, the person has a fasting
glucose test done ...

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Glucose - Wikipedia

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